Acta vet. scand. 1972, 13, 348-371.

From the Department of Physiology, Endocrinology and Blood Grouping, Royal Veterinary and Agricultural University, Copenhagen, Denmark.

PURIFICATION AND PROPERTIES OF THREE PIG

ERYTHROCYTE CARBONIC ANHYDRASES

By

Jørn Hejgaard

HEJGAARD, JØRN: Purification and properties of three pig erythrocyte carbonic anhydrases. Acta vet. scand. 1972, 13, 348—371. — Three distinct forms of the zinc containing enzyme carbonic anhydrase were isolated from pig erythrocytes. One low activity type enzyme and two genetic variants of the high activity type enzyme with identical CO₂ hydratase activities which were 8 times as high were isolated from Danish Black and White Swine. In the isolation procedure described, the hemoglobin was eliminated by precipitation with chloroform-ethanol, and the isoenzymes were separated by DEAE-Sephadex chromatography. A number of enzymatically active minor components were separated. They were apparently all genetically linked to one of the three major components. The three purified isoenzymes behaved as homogeneous components during isoelectric focusing and electrophoresis at different pH values. They were characterized in terms of molecular weight, isoelectric pH, zinc content, amino acid composition, and enzymatic activity against CO₂, p-nitrophenyl acetate, and β -napthyl acetate. The circular dichroism of the enzymes in the ultraviolet region was studied. The properties of the enzymes were similar to those of carbonic anhydrases of corresponding types isolated from other mammalian species. Sulphur containing amino acid composition of the two high activity type enzyme. The amino acid composition of the two high activity mutants deviated only in that an arginine residue in the most widespread genetic variant was replaced by a histidine residue in the less frequent variant. Otherwise the two mutants showed identical properties.

carbonic anhydrase; pig; erythrocyte; isoenzymes; genetic variants; DEAE-Sephadex; isoelectric pH; amino acid composition; CO_2 hydratase activity; esterase activity; circular dichroism.

Multiple molecular forms of the enzyme carbonic anhydrase have been isolated from red cells of many mammalian species including man, rhesus monkey, cow, horse, dog, guinea pig, rabbit and rat. The results of numerous studies of the chemical, physical, and enzymatic properties of the carbonic anhydrase isoenzymes and of their physiological role have been reviewed by *Edsall* (1968) and *Maren* (1967).

Carbonic anhydrases so far isolated can be classified into two chemically distinct enzyme types. Owing to their highly different enzymatic activity they have been designated "high activity type" and "low activity type" enzymes, respectively. Normally both molecular forms are present in the red cells, but only one type has been demonstrated in cattle. Presumably one type of the enzyme arose from the other by a process of gene duplication. Studies of isoenzymes in primates have shown that the two types of enzyme are under separate genetic control (Tashian 1969). Genetic variants of one or both types were observed in many species including man and other primates (Tashian), cattle (Sartore & Bernoco 1966), horse (Sandberg 1968), and sheep (Tucker et al. 1967). Only the rarely encountered variants of the human low activity enzyme have been studied to some extent. The observed polymorphisms are apparently all regulated by codominant autosomal allelic genes. A number of enzymatically active minor components encountered in apparently all species are undoubtedly conformational isomers which do not represent additional genetic loci (Tashian; Funakoshi & Deutsch 1969).

A study of the pig erythrocyte carbonic anhydrases was initiated by the demonstration in this laboratory of two genetic variants of the high activity enzyme in Danish Black and White Swine (*Kloster et al.* 1970). The frequency of the least widespread type was 20 % (homozygote plus heterozygote). One low activity type of enzyme was present in all examined animals.

The three isoenzymes were isolated in an apparently homogeneous form. Some physical and chemical properties of the two types of the enzyme and their genetic variants in this species were characterized to form the basis of studies of the distribution and function of the isoenzymes in pig tissues other than blood. It was also of interest to compare with carbonic anhydrases from other mammalian species at an equal evolutionary stage.

MATERIALS AND METHODS

Chemicals. The following reagents were used: Sephadex G-75 superfine and DEAE-Sephadex A-50 (Pharmacia, Uppsala, Sweden), acetazolamide (K & K Laboratories, Hollywood, Calif., USA); ovalbumin, bovine serum albumin (fract. V), p-nitrophenyl acetate and β -naphthyl acetate (Sigma, St. Louis, Mo., USA); porcine trypsin (Novo, Copenhagen, Denmark); Fast Blue BB Salt (diazotized 4-benzoylamino-2,5-dimethoxyaniline, ZnCl₂) (Gurr, London S.W. 14, England); Ampholine, pH 3—10 and pH 5—8 (LKB-Produkter, Bromma 1, Sweden). Danish potato starch (Struers, Copenhagen, Denmark) was hydrolyzed according to *Smithies* (1955). Most other chemicals were of analytical reagent grade.

Preparation of chloroform-ethanol extract. Freshly obtained blood was used in all experiments. The erythrocytes were isolated by centrifugation, and after removal of the supernatant and the leucocyte layer they were washed three times with 0.9 % NaCl. The packed cells were hemolyzed by addition of an equal volume of water and dialyzed overnight at 4°C. The extraction procedure applied was essentially that of Armstrong et al. (1966). After the removal of the denaturated hemoglobin by centrifugation the supernatant was filtered and dialyzed at 4°C.

Concentration of protein solution. The extract was concentrated by placing the dialysis tubes in solid $(NH_4)_2SO_4$. This procedure reduced the volume of the enzyme extract to one fifth of its original volume in less than two hrs. The tube was then transferred to a saturated $(NH_4)_2SO_4$ solution adjusted to pH 7.0 with solid Tris and left for 12 hrs. at 4°C. The precipitated protein was isolated by centrifugation, and the precipitate was stored at 0°C.

DEAE-Sephadex chromatography. Separation by column chromatography was in all cases performed at 7—8°C. Elutions were carried out with 0.045—0.2 M Tris adjusted with HCl to pH 8.2 at 25°C. The flow rate was 25—30 ml per hr., and fractions of 10.5 or 13.5 ml were collected. DEAE-Sephadex A-50 was pretreated according to the manufacturer's directions and the suspension was equilibrated with start buffer in a 25 \times 500 mm column. The protein concentrate was dialyzed against the 0.045 M start buffer, filtered, and applied to the top of the column. The low activity carbonic anhydrase was eluted with the start buffer, while the two other enzymes together with all the minor components were eluted by stepwise increase of the ionic strength of the buffer. The concentration and storage of the isolated fractions were performed according to the procedure described above. Some fractions were lyophilized for amino acid analysis, determination of nitrogen, etc.

Assay of CO_2 hydratase activity. In the course of the fractionations, the CO_2 hydratase activity was determined by a modification of the method of Wilbur & Anderson (1948). The incubation mixture was 1 ml of 0.025 M veronal buffer, pH 8.2, containing 1 % (w/v) Bromthymol Blue and 200 μ M EDTA. To this was added 0.1 ml of enzyme solution in 0.01 % peptone, and the solution was equilibrated at 0.5°C. Then 1 ml of H₂O saturated with CO₂ at the same temperature was injected. The time interval from injection until colour change from a yellowish-green to pure yellow was recorded and the specific activity calculated as described by *Rickli et al.* (1964). Conditions were so selected that the non-catalyzed reaction lasted 100—110 sec. and the catalyzed reaction 18—25 sec. Within this range, results were reproducible and the activity varied linearly with protein concentration.

Assay of esterase activity towards p-nitrophenyl acetate. The buffer, 2.4 ml of 0.0125 M Tris-maleate adjusted to pH 7.15 with NaOH, was equilibrated in a 1 cm cuvette at 30°C. Then 0.3 ml of a 12.5 mM solution of p-nitrophenyl acetate in 40 % (v/v) acetonitrile or in 50 % acetone was added. Absorbance at 400 nm was followed in a recording spectrophotometer through 9 min., using 0.01 M buffer as reference. After incubation for 2 min., 0.3 ml enzyme solution was added. After further 5 min., 0.1 ml of 4 mM acetazolamide was added, upon which the rate of increase in absorbance dropped to the non-enzymic level. The esterase activity was determined as the difference between rates of hydrolysis after and before enzyme had been added. The readings before addition of enzyme and after addition of acetazolamide provided an internal blank and enabled a correction for contributions from other esterases present in impure preparations. In kinetic assays the substrate concentration was varied in the range 0.2—2 mM.

Assay of esterase activity towards β -naphthyl acetate. An enzyme solution in 1.6 ml of water was added to 0.2 ml of 0.1 M Tris-maleate buffer adjusted to pH 7.15 with NaOH, and the mixture was equilibrated at 30°C. Subsequently, 0.2 ml of a 12.5 mM β -naphthyl acetate in 40% (v/v) acetonitrile or in 50% acetone was added, and the mixture was incubated for 20 min. at 30°C. The reaction was stopped by addition of 0.5 ml of 4 mM acetazolamide. Immediately after 0.5 ml of a 0.4% solution of Fast Blue BB Salt was added. The colour was developed for 40 min. at room temperature. Upon thorough mixing, 1 ml was withdrawn and diluted with 3 ml of 96% ethanol. The absorbance was measured at 550 nm against a reagent blank.

Starch gel electrophoresis was carried out according to the method of Smithies. The concentration of starch was 12 %. The gel and vessel buffer was 0.05 M Tris, adjusted with glycine to pH 8.8 and 8.4, respectively. A buffer system described by Kloster et al. (1970) was used in some experiments. Electrophoresis was carried out in a horizontal, water-cooled apparatus at 18 v/cm. The separated proteins were localized with Amido Black. Staining for esterase activity was carried out by placing the gel in an aqueous solution of 0.02 % β -naphthyl acetate, 0.2 % Fast Blue BB Salt, 5 % ethanol at 37°C for 3 hrs. Zones representing CO₂ hydratase activity were in some experiments localized by means of Bromthymol Blue and CO₂ (Tashian 1969).

Isoelectric focusing was performed in a 110 ml capacity apparatus (LKB-Produkter, Bromma 1, Sweden) as described by Vesterberg & Svensson 1966). The column contained 0.75 % (w/v) Ampholine pH 5-8, and 0.25 % Ampholine pH 3--10. It was stabilized against convection by a 46-0 % (w/v) continuous sucrose gradient. The voltage

was gradually increased from 200 to 400 v. The temperature of the cooling water was 10°C. When equilibrium was reached, fractions of 1.25 ml were collected and their pH determined immediately using a combination electrode connected to a Model 27 pH-meter (Radiometer, Copenhagen, Denmark). Protein content and enzyme activity were determined as in fractions isolated by column chromatography.

Purity of enzyme fractions was controlled by isoelectric focusing in polyacrylamide gels (*Wrigley* 1968) using 6×125 mm tubes. Chemical polymerization was used for the preparation of 6.5 % gels containing 2 % (w/v) of a mixture of equal volumes of Ampholine pH 3-10 and pH 5-8. Separation was done at 2 mA per tube until the voltage had risen to 300 v, at which level it was kept constant. Proteins were fixed in 12.5 % trichloroacetic acid and stained with Coomassie Blue (*Chrambach et al.* 1967).

Determination of molecular weight by gel filtration was carried out according to the method of Andrews (1964). Sephadex G-75 superfine was packed in a 1×120 cm column. The proteins were eluted at 2°C with 0.025 M phosphate buffer, pH 7.0, containing 0.25 M-KCl. The column was calibrated with bovine serum albumin, ovalbumin, and porcine trypsin. The elution volumes were determined by monitoring the effluent at 280 nm and, in the case of carbonic anhydrases, by assay of the CO₂ hydratase activity. A Blue Dextran reference was included in all experiments.

Amino acid analyses were performed in accordance with Spackman et al. (1958) using a Beckman 120 C amino acid analyzer provided with an Infotronics two-channel integrator. Periods of hydrolysis were 16, 24, and 48 hrs. The amounts of NH_3 , serine, and threonine were determined by extrapolation to zero time. The values obtained after 48 hrs.' hydrolysis were used for valine and isoleucine.

Tryptophan was determined using the same procedure except that the acid standard hydrolysis was supplied with 3 % (v/v) thioglycolic acid (*Matsubara & Sasaki* 1969). The samples were hydrolyzed for 24 hrs. Tryptophan values were calculated relative to contents of arginine and lysine, assuming 83 % recovery. In some control analyses of lysozymes from man and egg white, the recovery varied between 81 and 84 %. Tyrosine and tryptophan were also determined spectrophotometrically according to Goodwin & Morton (1946).

Half-cystine was determined as cysteic acid (Spencer & Wold 1969) using 24 hrs. hydrolysis in the presence of 2.5 % (v/v) dimethyl sulphoxide. Upon amino acid analysis, the content of half-cystine was calculated relative to aspartic acid, using the integration constant of the latter for quantitation of both amino acids. This method gave a 100 % recovery by analysis of human lysozyme and glutathione.

Determination of zinc. Enzyme solutions were dialyzed for 48 hrs. against three shifts of Zn-free water. The protein was precipitated in boiling 5 % TCA. Upon centrifugation, the zinc in the supernatant was determined with an EEL Model 140 atomic absorption spectro-photometer using the third outer dialysate as reference.

Determination of nitrogen. The micro Kjeldahl method of Parnas (1938) was employed.

Determination of protein. Absorbance at 280 nm served as a measure of protein concentration, taking $A_{280}^{1\%}$ as 18.6 for enzyme Ia and 18.8 for IIb and IIa. These values were calculated on the basis of amino acid analyses of enzyme solutions as described by *Rickli et al.* Calculations were made on the assumption that 1 mole of protein equalled 39 moles of basic amino acids (Lys + His + Arg) in all three proteins. The molecular weights used are recorded at the bottom of Table 2.

Circular dichroism was determined on a Roussel-Jouan Dichrograph CD 185. The spectra in the range 300-240 nm were recorded using a 1 cm cell and in the 240-200 nm range using a cell with 0.2 cm light path. The Dichrograph was operated at maximal sensitivity. Mean residue molecular weights obtained from the amino acid analysis data were used in calculating the mean residue molecular ellipticity (θ).

RESULTS

Nomenclature. The nomenclature used by Tashian (1969) in his classification of primate carbonic anhydrases was preferred to the generally accepted B/C-nomenclature (Rickli et al. 1964, Furth 1968). Tashian's nomenclature provides a distinct division in two enzyme types and it permits an unambiguous subclassification of the genetic variants of the two types and their conformer minor components. The low activity type enzyme isolated was designated Ia, the two high activity mutants designated IIa and IIb, respectively, in conformity with the A/B-nomenclature used by Kloster et al. (1970). Peaks from ion exchange or electrophoretic bands representing conformational isomer variants of the three major components, or apparently well-defined enzyme aggregates, are designated + or -, according as the mobility in DEAE-Sephadex, or electrophoretically towards the anode, is higher or lower than that of the corresponding major component $(e.g., Ia^+, IIb^-).$

Preparation of carbonic anhydrase isoenzymes. Extraction with chloroform-ethanol resulted in an almost 100 % precipitation of the hemoglobin. The extracts were of a yellowish colour, and the protein precipitated with $(NH_4)_2SO_4$ was pale greyish. More than 65 % of the enzyme activity of the hemolysate was recovered in this concentrate. No further activity could be iso-

lated from the hemoglobin paste. The carbonic anhydrases could also be separated from the hemoglobin by DEAE-Sephadex chromatography of the dialyzed hemolysate (Armstrong et al. 1966). Attempts to isolate the hemoglobin by precipitation with $(NH_4)_2SO_4$ (Furth) gave unsatisfactory results.

The three isoenzymes were separated by DEAE-Sephadex chromatography in a single run. The low activity enzyme Ia and the high activity type IIb were isolated from hemolysates from animals which, regarding the high activity type enzyme, were homozygous. The IIa form was isolated together with Ia and IIb from blood derived from heterozygotes. Separation was optimal when Ia was eluted with 0.05 M, IIb with 0.1 M, and IIa with 0.15 M buffers. Gradient elution always resulted in an incomplete separation of major and minor components. Fig. 1 shows a typical result of DEAE-Sephadex chromatography. For clarity, only the hydratase activity of the three major components in the extract is recorded. Several other protein peaks were observed, some of which showed CO₂ hydratase activity.



F i g u r e 1. Elution of Ia, IIb, and IIa isoenzymes from DEAE-Sephadex. (-----) absorbance at 280 nm; (-----) CO₂ hydratase activity. The enzymes were eluted with Tris-HCl pH 8.2 (25°C), Ia with 0.05 M, IIb with 0.1 M, and IIa with 0.175 M buffer as indicated at the top of the figure. The left ordinate scale applies to fractions 1-43, while the remaining fractions refer to the right hand scale. Absorbance was measured after appropriate dilution. Volumes of the collected fractions were 13.5 ml.

Electrophoresis of chloroform-ethanol extracts from larger numbers of pigs revealed the presence of a considerable number of active minor components. Animals of all three high activity enzyme phenotypes were studied. Both of the major high activity components were always associated with at least 4 active minor bands which, undoubtedly, are products of the same genetic locus. The bands were located in identical positions relative to the major bands, and corresponding bands probably represent the same conformational changes in their respective major forms. Together with the low activity enzyme band Ia, two very weak bands with mobilities between those of Ia and IIb could be demonstrated in all extracts. These bands are probably analogous variants of the Ia enzyme.

This feature is illustrated by the results obtained by DEAE-Sephadex chromatography of a chloroform-ethanol extract with an atypically high content of minor components (Fig. 2). The extract was derived from a homozygous animal (type IIb). Three protein peaks without CO_2 hydratase activity were first eluted. After elution of the low activity enzyme (Ia), two diminutive peaks with low activity (Ia⁺) appeared. Then at least three peaks with high activity (IIb⁻) followed, prior to elution of the major component IIb. Finally, one or two high activity carbonic anhydrases were eluted (IIb⁺) with 0.2 M buffer. It is obvious that



F i g u r e 2. Elution of conformere Ia and IIb isoenzymes from DEAE-Sephadex. (_____) absorbance at 280 nm; (----) CO₂ hydratase activity. Concentrations of elution buffers are indicated at the top of the figure. 10.5 ml fractions were collected.

the high activity enzyme is by far the most labile type. In this extract, the minor components represented more than 60 % of the high activity protein. The CO_2 hydratase activity of the minor components was always lower than that of the major component.

The minor form present in greatest quantity was designated IIb⁻¹. This component could be identified in chloroform-ethanol extracts as a well-defined chromatographic peak, and in electrophoresis it migrated apparently as one band. A complete purification was not attempted. One fraction collected from the central part of the IIb⁻¹ peak (Fig. 2) was examined. The CO, hydratase activity and the esterase activity were 75-80 % of that of the IIb component. The fraction was concentrated by $(NH_4)_{2}SO_{4}$ and lyophilized. Electrophoresis of this product showed that more than 80 % of the protein was by now present in the IIb band, the remaining percentage being localized in two enzymatically active, weak bands of lower anodic mobility. The mobility of the IIb and Ha forms were not changed by lyophilization. It may thus be assumed that the IIb⁻¹ peak (Fig. 2) represents a IIb variant, generated by a reversible conformational change. A corresponding IIa⁻¹ form was observed. The intensity of the secondary bands varied much from extract to extract, depending on minor variations in the procedure. In some experiments where extremely high concentrations of chloroform-ethanol were used for the extraction, a subsequent electrophoresis showed that IIb⁻¹ and IIa⁻¹ were the predominating high activity enzyme bands present. Staining with Bromthymol Blue, however, revealed the presence of the high activity minor bands also in crude hemolysates. Thus, they are to some extent present in vivo or might be formed during electrophoretic separation.

Table 1 gives typical examples of composition of chloroformethanol extracts, as calculated after DEAE-Sephadex chromato-

	II	Carbonic anhydrase		
	phenotype	Ia	IIb	IIa
Per cent of	bb	76	24	
active protein	ab	77.5	11.5	11
Per cent of CO ₂	bb	28	72	
hydratase activity	ab	29.5	36	34.5

Table 1. Separation of crude carbonic anhydrase preparations in genetic variants by DEAE-Sephadex chromatography.



Figure 3. Starch gel electrophoresis (top) and isoelectric focusing (bottom) of the fractions isolated after rechromatography on DEAE-Sephadex. Approx. 80 µg of the isolated isoenzymes were applied. (CEE) chloroform-ethanol extract; (MIX) mixture of the three isoenzymes; (A) application of sample; (⊕) anode. Some minor components in the chloroform-ethanol extract are indicated in the left side of the figure. To the right is shown an approximate pH scale applying to the isoelectric focusing.

graphy. At least 90 % of the total activity applied to the column was recovered. Contributions from minor components were added to the value of the major component from which they probably were formed and ascribed the same specific activity as the latter. Within experimental error, the distribution between high and low activity carbonic anhydrases was identical in heterozygous individuals and homozygous animals of the examined type. In extracts of blood from heterozygotes the ratio between the amounts of the two high activity enzymes was very close to unity. In studies of genetic variants of the human low activity carbonic anhydrase, *Tashian* observed that the three rare mutants included in his study were produced in lower quantities than the normal enzyme.

Minor components and other proteins, if any, in pooled fractions from the major peaks from several runs were eliminated by rechromatography. The proteins were eluted with buffers with lower ionic strength than previously applied, but otherwise under identical conditions. Approximately 5 times the bed volume of buffer was used. All three isoenzymes appeared in fully overlapping, symmetrical protein and activity peaks. Electrophoresis at pH 7 and 8.8 of fractions from rechromatography revealed in all three cases apparently homogeneous proteins (Fig. 3). The difference in electrophoretic mobility of the IIb and IIa forms increased with pH. After isoelectric focusing, the three fractions stained as distinct single bands. This criterium, however, is less conclusive, since the two high activity enzymes have identical isoelectric pH, as illustrated in Fig. 3. In Sephadex gel filtration experiments the three preparations were eluted as single, symmetric peaks. Amino acid analysis of the low activity enzyme showed that cysteine and methionine were absent in this protein. The total absence of these two peaks in the chromatograms confirms the purity of the isolated Ia protein.

Properties of the isoenzymes

Ultraviolet absorption. The molar extinction coefficient ε_{M} at 280 nm, which was used as standard measure of protein concentration, was calculated on the basis of the amino acid analyses as described under "Methods". The values obtained were 53,700 M⁻¹ cm⁻¹ for Ia and 54,000 M⁻¹ cm⁻¹ for IIb and IIa. ε_{M} was also calculated after Kjeldahl nitrogen analyses of protein solutions

with measured absorbance. Nitrogen percentages calculated from the amino acid analyses were used. The values obtained by this method were 2-3 % lower than those presented above. The reason may be that the amide content in the proteins was overestimated. The uncertainty involved in the amide determination was marked, and the values recorded in Table 2 are higher than those observed in carbonic anhydrases from other species. Extinction coefficients may also be obtained by summation of contributions from the individual tyrosine and tryptophan residues. Values calculated in this way were in all three proteins 13 % lower than those observed. This discrepancy is of the same magnitude as observed in a series of other proteins (*Wetlaufer* 1962).

Molecular weight. Gel filtration of the three isoenzymes resulted in all experiments in molecular weights ranging between 30,000 and 30,500. Thus, molecular weights of the three carbonic anhydrases are assumed to be nearly identical. It is recognized that the elution volumes are more directly determined by Stokes' radii, and the absolute value obtained is probably more uncertain.

Zinc content. Zinc analyses of the three enzymes gave for Ia, 0.227 %; for IIb, 0.228 %; and for IIa, 0.235 %. These results represent the average of three analyses, and the maximal deviation was less than \pm 0.010 %. On the assumption that each molecule contains one Zn atom, the results indicate that the molecular weights of the three proteins are close to 29,000.

The amino acid composition of the three isolated isoenzymes is presented in Table 2. Initially the composition was calculated on the assumption that all three proteins contain 260 residues per molecule. Optimal adjustment to integral figures in the range 245—270 residues per molecule was obtained for Ia when the value of aspartic acid was assumed to be exactly 33.0 residues per molecule. For both high activity enzymes this was achieved, assuming 29.0 glutamic acid residues per molecule. The results obtained in this way are recorded to the left in the table. The values were then normalized to the nearest integer. The calculated molecular weights were for all three proteins found to be close to 28,800, assuming one Zn atom per molecule. They are well within the range found by Zn determination and gel filtration.

Tyrosine and tryptophan were also determined spectrophotometrically. When this method is applied, it is essential to await Table 2. Amino acid composition of pig carbonic anhydrase Ia, IIb and IIa.

The calculated molecular weights equal the sum of the amino acid residues plus 1 zinc plus 1 H_2O . The nitrogen recovery was calculated on the basis of Kjeldahl analysis of the sample used for amino acid analysis.

	Residues per molecule						
		found			to nearest integer		
	Ia	11b	IIa	Ia	IIb	IIa	
Lysine	21.2	17.6	17.9	21	18	18	
Histidine	12.9	12.1	13.3	13	12	13	
Arginine	4.9	8.6	7.6	5	9	8	
Amide NH,	(33)	(35)	(35)	(33)	(35)	(35)	
Aspartic acid	33.0	27.3	27.2	33	27	27	
Glutamic acid	20.3	29.0	29.0	20	29	29	
Glycine	17.7	22.8	23.1	18	23	23	
Alanine	19.4	12.0	12.0	19	12	12	
Valine	15.0	15.7	15.6	15	16	16	
Leucine	22.0	21.3	21.0	22	21	21	
Isoleucine	15.7	9.5	9.5	16	10	10	
Proline ^a)	17.0	17.5	17.5	17	17	17	
Serine	26.8	19.3	18.9	27	19	19	
Threonine	8.7	12.9	12.9	9	13	13	
Half-cystine ^b)	0.0	2.1	2.1	0	2	2	
Methionine	0.0	2.6	2.4	0	3	3	
Phenylalanine	9.0	11.5	11.6	9	12	12	
Tyrosine	10.2	6.0	6.0	10	6	6	
Tryptophan ^c)	5.8	7.0	7.0	6	7	7	
Total no. of residues					256	256	
Nitrogen recovere	d, %			99	101	98	
Calculated molecular weight					28802	28783	

^a) The observed yields were expected to be slightly high when cysteine is present.

b) Determined as cysteic acid by the method of Spencer & Wold (1969).

^c) Determined by the method of Matsubara & Sasaki (1969).

a complete ionization of the tyrosine residues (*Beaven & Holiday* 1952). The low activity carbonic anhydrase was apparently unfolded after a few min. at pH 13. Absorption spectra of solutions of the two high activity type enzymes in 0.1 M-NaOH changed steadily during three hrs. at room temperature after which the absorbance at 280 and 294.4 nm became constant.

360

Absorption measurements after alkaline denaturation for $3\frac{1}{2}$ hrs. gave tryptophan values in fair agreement with those obtained by amino acid analyzer, namely 5.8, 7.1, and 7.2, residues per molecule of Ia, IIb, and IIa, respectively. The observed tyrosine values, however, were higher than those obtained by the amino acid analyzer, and the ratio tyrosine to tryptophan was 2.10 for Ia and 1.25 for the two high activity enzymes. If these values were used with the tyrosine values obtained by amino acid analyses for a calculation of the tryptophan content, only 5 residues should be present per molecule of all three proteins. A tryptophan content of this magnitude can hardly be in agreement with the observed molar extinction coefficients. The reason for this discrepancy may be the protracted equilibration period preceding the photometry.

An interesting feature of the amino acid analyses is that sulphur is completely absent in the low activity type enzyme. All carbonic anhydrases so far analysed have been found to contain at least one methionine residue. Absence of cysteine was observed only in the high activity bovine enzyme (cf. *Furth*).



F i g u r e 4. Isoelectric focusing of a mixture of isoenzymes from ion exchange chromatography. Ia (approx. 220 μ g); IIb (approx. 65 μ g); IIa (approx. 80 μ g). (-----) ΔA_{280} (absorbance of effluent fraction minus calculated contributions from carrier ampholytes); (-----) CO₂ hydratase activity; (•) pH. Absorbance at 280 nm was measured directly. As the Ampholine represents a relatively large proportion of the measured values, ΔA_{280} provides only an approximate expression of the protein concentration. Neither protein nor enzymatic activity were demonstrable outside the depicted pH-range. The collected fractions were 1.25 ml.

J. Hejgaard

Within the experimental error, the amino acid analyses of the two high activity carbonic anhydrases resulted in exactly the same values for all the amino acids except two. IIb was found to contain 12 histidine and 9 arginine, while the more rare mutant IIa contained 13 histidine and 8 arginine. This difference was substantiated by analyses of various enzyme preparations. As to the other amino acids the difference was maximally 2 % of the observed number of residues, or maximally 0.3 residues.

Isoelectric pH values were determined by focusing of small amounts of the three proteins in stable Ampholine gradients, as illustrated in Fig. 4. The isoelectric pH values thus obtained were 7.15 for Ia and 5.97 for IIb and IIa. The maximum deviation from these values was ± 0.05 pH units. As to the two high activity enzymes it proved impossible to demonstrate any difference in isoelectric pH (cf. also Fig. 3). The enzymatic activity of all three proteins remained unchanged after isoelectric focusing.

Enzymatic activity. Table 3 shows the enzymatic activities of three isoenzyme preparations towards the substrates CO_2 , p-nitrophenyl acetate, and β -naphthyl acetate. In accordance with findings in other species, the catalytic properties of the two types of enzyme were highly different, but significant differences in specific activity of the two high activity forms were not demonstrable. When the CO_2 hydratase activity was assayed according to Wilbur & Anderson's method, the type II enzymes were found to be more than 8 times as active as the Ia enzyme. Assays of the relatively weak esterase activity against p-nitrophenyl acetate

Table 3. CO_2 hydratase and esterase activities of pig carbonic anhydrase Ia, IIb and IIa.

The CO, hydratase	activity is	measured	in Will	our-Anderson	units
(Rickli et al. 1964).	The units	of the ester	ase activ	ities are µmo	oles of
phenolic compound	formed per	r minute pe	er mg pr	otein. The es	terase
assays were perform	rmed in 4 9	% (v/v) ac	etonitrile	e or 5 % acet	o ne.

Carbonic anhydrase	CO ₂	Esterase activity					
	hydratase	p-nitropheny	yl acetate	β-naphtyl acetate			
	activity	acetonitrile	acetone	acetonitrile	acetone		
Ia	29 000	0.15	0.13	0.133	0.095		
IIb	$240\ 000$	1.41	1.54	0.0165	0.0178		
IIa	$245\ 000$	1.50	1.56	0.0169	0.0181		

362

showed an activity ratio of the same order of magnitude. With β -naphthyl acetate as substrate, the picture was reversed, and the catalytic capacity of the low activity enzyme Ia was the higher one.

In order to reproduce the CO₂ hydratase assays it was essential to supply EDTA to the mixture. Type II enzymes, which were present in very low concentrations when assayed, gave very fluctuating specific activities when EDTA was not added. It is assumed that the enzymatic activity is affected by traces of metal ions that may be present in the assay mixture or become bound to the protein during isolation. When the activity of type II enzymes was assayed in the presence of 1.5 µM Cu++, but in the absence of EDTA, a 80 % inhibition was observed. A Cu++ inhibition of the same relative magnitude has been found with the human high activity enzyme (Magid 1967). The activity towards β -naphthyl acetate depended, as apparent from the table, on the solvent used. If assayed in 4 % acetonitrile or in 5 % ethanol, the specific activity ratio Ia to II was 8, while it was approx. 5 in 5 % acetone. The solvent effect on the p-nitrophenyl acetate esterase assays was less pronounced. None of the enzymes were saturated with substrate at the maximum attainable concentration of p-nitrophenyl acetate. The initial velocity of the reaction was almost proportional to the substrate concentration under these conditions, and no satisfactory determination of kinetic constants could be made. Similar results have been obtained with enzymes from other species (McIntosh 1969, 1970).

All activity of the three enzyme preparations was completely inhibited by 0.2 mM acetazolamide. Thus, the enzymes did not contain demonstrable amounts of other esterases which are not affected by this potent carbonic anhydrase inhibitor.

Circular dichroism (CD). Carbonic anhydrases produce complex, but characteristic CD spectra at wave lengths in the range 210—300 nm. Spectra of low and high activity enzymes isolated from human erythrocytes presented considerable differences, particularly in the range 260—220 nm (Beychok et al. 1966). Fig. 5 shows CD spectra of the three isolated pig enzymes in native form at pH 7. Within experimental error the two high activity carbonic anhydrases IIa and IIb produced completely identical spectra. The CD spectra of corresponding high and low activity enzymes derived from man and pig contained almost



Figure 5. CD spectra of carbonic anhydrase isoenzymes in the range 300-200 nm in 0.025 M phosphate buffer, pH 7.0. (-----) Ia; (-----) Ib and IIa. The dotted parts are subject to great uncertainty owing to the high noise level. The left ordinate applies to the 200-240 nm range. The scale to the right applies to the 240-300 nm range.

the same characteristic major bands, but the spectra presented distinct differences in the finer details.

The two high activity enzymes produced a negative extreme at 268 nm. The spectrum became positive at 255 nm, and at 244 nm a distinct maximum was found. In the range from 236 to 205 nm, or below, the spectrum was negative with a maximal negative ellipticity near 209 nm. Similar to this, the low activity enzyme presented a negative band centered close to 275 nm, a very broad positive band with maximum ellipticity at 234 nm, and a minimum at approx. 212 nm. The finer structures of the two spectra showed many similarities. In the range 300-280 nm, small negative extremes were demonstrable at 293 nm and at 285 nm, and maxima were observed at 289 nm and 281-2 nm, and at 272 nm in the II spectra. After passage of the minimum, the Ia spectrum showed a shoulder at 270 nm, and in some spectra a slight shoulder was observed at 250 nm. At wavelengths lower than the extreme positive maximum, a well-defined positive band was seen in the Ia spectrum at 225 nm and, at approximately the same wavelength, a distinct shoulder was present in the II spectrum. Finally, spectra of the low activity enzyme presented a shoulder at 220 nm, and some spectra of the IIb enzyme indicated a shoulder at 218 nm. Below 208 nm spectra were recorded at a very high noise level, and, consequently, the dotted part of the curves should be taken with reservation. Spectra of lyophilized enzyme preparations did not deviate significantly from those of native enzymes.

CD spectra are sensitive registrators of conformational changes in protein molecules. The slow denaturation of the Ia and IIb enzymes under moderately acid and alkaline conditions was studied to some extent. After incubation for 3 hrs. at 25°C in 0.025 M formate buffer, pH 4, the finer details of the spectrum of the high activity enzyme remained unchanged in the range 300-245 nm, but the ellipticity in the negative extreme was reduced by approx. 35 %. At lower wavelengths, drastic changes had occurred in the spectrum of the enzyme. The distinct band at 345 nm had extended towards lower wavelengths in a broad, flat peak, and the cross-over point, at 236 nm, at pH 7, had moved downward to 223 nm. Any significant differences were not observed in the vicinity of the minimum point at 209 nm. Changes in the spectrum of the low activity enzyme were not equally marked. The most characteristic finding was a reduced ellipticity at the positive maximum at 225 nm. After incubation for 3 hrs. at pH 4, the two enzyme types thus had almost identical crossover points from positive to negative ellipticity at approx. 223 nm. After incubation for 90 min. at 25°C in 0.025 M carbonate buffer, pH 10.0, the IIb spectrum was apparently identical with those recorded at pH 7. The spectrum of the low activity enzyme was changing in a way similar to that observed in studies of the corresponding human enzyme at high pH (Beychok et al.). Most characteristic was a change to more positive ellipticity values in the range 240-287 nm, while the characteristic, finer structures between 267 and 287 nm apparently remained intact.

DISCUSSION

The general properties of the pig carbonic anhydrases were very similar to those of the corresponding enzymes from other mammalian species. The isoenzymes unambiguously could be classified as low activity type or high activity type enzymes, respectively, on the basis of the structural differences reflected in their amino acid composition, enzymatic activity, and circular dichroism.

As observed in man, monkey, horse, and other species, the

low acticity type enzyme was predominating. The high activity type seemed to be more labile and, depending on methods used for isolation, up to 60 % of each of the two high activity enzyme types could be isolated as minor components, all of which showed an enzymatic activity lower than that of the corresponding major enzyme peak. Similar results have been obtained in a study of the horse high activity enzyme (*Furth* 1968). It has been proposed that the minor forms could be generated from the major ones by hydrolysis of amide groups (*Funakoshi & Deutsch* 1969). The regeneration of the IIb form from the isolated IIb⁻¹ fraction during concentration or lyophilization apparently excludes this explanation. It seems more probable that the relatively harsh procedure of lyophilization may have reverted a conformational change which originally had been produced by the chloroformethanol extraction or by other steps in the isolation procedure.

A comparison with the amino acid composition of the corresponding enzymes isolated from man, monkey, cattle, and horse (data may be obtained from Furth) supports the suggestion of Furth that the higher serine content in the low activity than in the high activity enzymes is the most characteristic difference in the composition of the two types. The low activity enzyme of pigs differs from that of other species so far studied in a relatively high content of isoleucine and the absence of sulphur containing residues. The values obtained do not provide a basis for extensive taxonomic considerations. The content of lysine in the high activity enzymes of pigs and in the corresponding enzymes of cattle and horse is relatively low as compared with the primate enzymes. Conversely, the contents of alanine and isoleucine, which almost equal the values from man and monkey, differ characteristically from those observed in the two other ungulate species.

The amino acid analyses indicate that one genetic variant of the high activity enzyme has been produced from the other by a single amino acid replacement, which can be due to a change of one nucleotide in the codon. The amino acids involved are arginine and histidine. The results of the amino acid analyses seem to be substantiated by the fact that the two proteins apparently have identical isoelectric points below pH 6, where arginine and most of the histidine are positively charged, and an increasing difference in relative electrophoretic mobility from pH 7 to 8, where the histidine residues become more and more uncharged, while the arginine side chains remain protonized. The residue involved is thus assumed to be at or near the protein surface. The uniform, systematic distribution of the minor components of the two enzymes, as observed by electrophoresis, indicates that this position in the molecule is not involved in processes resulting in a generation of minor components. All other investigated properties of the two high activity enzymes were very similar. Apparently the two mutants were synthesized in identical quantities in the examined heterozygous individuals. Whether the observed difference in composition is the only one cannot be decided until further data are available.

The recorded enzymatic activities show the same characteristic differences between high and low activity enzymes as have been observed in other species. The marked difference in esterase activity towards p-nitrophenyl acetate and β-naphthyl acetate, respectively, seems to be a characteristic feature of the presence of two types of enzymes. In man (Tashian 1969), pig, rabbit (McIntosh 1970), and rat (McIntosh 1969), activities of the high activity enzymes towards p-nitrophenyl acetate ranged very constantly from 5 to 10 times that of the low activity enzymes from the same species. The corresponding ratios between CO₂ hydratase activities were of the same order of magnitude, but varied more depending on the methods applied. In contrast, the low activity type enzyme in the 4 species showed in all cases a higher activity towards β -naphthyl acetate. Also the marked solvent effect on the latter activity of especially Ia indicates differences in the catalytic mechanism for hydrolysis of p-nitrophenyl acetate and β -naphthyl acetate, respectively, of at least one of the enzyme types.

The circular dichroism spectra contain the same major ellipticity bands as those observed in the two corresponding human enzymes. A correlation with ORD studies of carbonic anhydrases of horse (*Furth*) and primates (*Beychok et al.* 1966, *Rosenberg* 1966) confirms that the characteristic difference in structure of the two types of enzyme is reflected in the optical asymetry in their absorption bands. The general features of the very similar human enzyme CD spectra have been discussed in detail by *Beychok et al.* The CD bands in the wavelength range studied are attributable primarily to optically asymmetric interactions of the aromatic amino acids, in particular tyrosine and tryptophan. This may explain the observed differences in the fine structure patterns of the spectra of the corresponding enzymes from man and pig, and, correspondingly, the different effect of a moderate raise of pH on the pig enzymes may depict the greater content of tyrosine residues in the Ia enzyme. A more detailed discussion of these features seems rather futile until more information of the three-dimensional structures of the two types of enzymes is available. The band at 222 nm which is characteristic of the right handed α -helix was present in spectra of neither human nor pig enzymes. The few spectra recorded at pH 4 and the absorption measurements at pH 13 indicate that the high activity pig enzymes are more sensitive to acid denaturation than the low activity enzyme, and, on the other hand, the high activity enzyme apparently is more stable at the high pH. The reverse seems to apply to the human enzymes (*Beychok et al.*).

ADDENDUM

When preparation of the present manuscript was essentially terminated, two studies of porcine erythrocyte carbonic anhydrases were published. *Tanis et al.* (1970) have isolated and in part characterized one low activity and one high activity enzyme form. A study by *Ashworth et al.* (1971) was concentrated on a low activity form. Findings in these studies and the results presented in this paper supplement and confirm each other, but a comparison unveils some interesting differences which will be discussed in some detail elsewhere (*Hejgaard* 1971).

Based on a comparison of the recorded amino acid compositions it is concluded that the same molecular form of the low activity type enzyme was studied by all three workers. Either of the two high activity enzyme preparations from Danish pigs showed a composition identical with that of the high activity isoenzyme isolated by *Tanis* et al.

ACKNOWLEDGEMENTS

The author is indebted to Professor J. Moustgaard who initiated these investigations. The excellent assistance of Mr. E. Lund, The Danish Institute of Protein Chemistry, with amino acid analyses is gratefully acknowledged.

REFERENCES

- Andrews, P.: Estimation of the molecular weights of proteins by Sephadex gel filtration. Biochem. J. 1964, 91, 222-233.
- Armstrong, J. McD., D. V. Myers, J. A. Verpoorte & J. T. Edsall: Purification and properties of human erythrocyte carbonic anhydrases. J. biol. Chem. 1966, 241, 5137-5149.

- Ashworth, R. B., T. E. Spencer & J. M. Brewer: Purification and properties of carbonic anhydrase B from porcine erythrocytes. Arch. Biochem. 1971, 142, 122-131.
- Beaven, G. H. & E. R. Holiday: Ultraviolet absorption spectra of proteins and amino acids. Advanc. Protein Chem. 1952, 7, 319-386.
- Beychok, S., J. McD. Armstrong, C. Lindblow & J. T. Edsall: Optical rotatory dispersion and circular dichroism of human carbonic anhydrases B and C. J. biol. Chem. 1966, 241, 5150-5160.
- Chrambach, R., R. A. Reisfeld, M. Wyckoff & J. Zacchari: A procedure for rapid and sensitive staining of protein fractionated by polycrylamide gel electrophoresis. Analyt. Biochem. 1967, 20, 150-154.
- Edsall, J. T.: The carbonic anhydrases of erythrocytes. Harvey Lect. Ser. 1968, 62, 191-230.
- Funakoshi, S. & H. F. Deutsch: Human carbonic anhydrases. II. Some physicochemical properties of native isoenzymes and of similar isoenzymes generated in vitro. J. biol. Chem. 1969, 244, 3438-3446.
- Furth, A. J.: Purification and properties of horse erythrocyte carbonic anhydrases. J. biol. Chem. 1968, 243, 4832-4841.
- Goodwin, T. W. & R. A. Morton: The spectrophotometric determination of tyrosine and tryptophan in proteins. Biochem. J. 1946, 40, 628-632.
- Hejgaard, J.: Studier over enzymer hos svin. III. Svinets erytrocytære carboanhydraser. (Studies on enzymes in pigs. III. Pig erythrocyte carbonic anhydrases). Kgl. Vet.- og Landbohøjsk., Inst. Sterilitetsforskn. Aarsberetn. 1971, 153-168. København.
- Kloster, G., B. Larsen & P. B. Nielsen: Carbonic anhydrase polymorphism in cattle and swine. Acta vet. scand. 1970, 11, 318-321.
- Magid, E.: The activity of carbonic anhydrases B and C from human erythrocytes and the inhibition of the enzyme by copper. Scand. J. Haemat. 1967, 4, 257-270.
- Maren, T. H.: Carbonic anhydrase: chemistry, physiology, and inhibition. Physiol. Rev. 1967, 47, 595-781.
- Matsubara, H. & R. M. Sasaki: High recovery of tryptophan from acid hydrolysates of proteins. Biochem. biophys. Res. Commun. 1969, 35, 175—181.
- McIntosh, J. E. A.: Carbonic anhydrase isoenzymes in the erythrocytes and dorsolateral prostate of the rat. Biochem. J. 1969, 114, 463-476.
- McIntosh, J. E. A.: Carbonic anhydrase isoenzymes in the erythrocytes and uterus of the rabbit. Biochem. J. 1970, 120, 299-310.
- Parnas, J. K.: Über dis Ausführung der Stickstoffbestimmung nach Kjeldahl in der Modifikation von Parnas und Wagner. (The Kjeldahl method for determining nitrogen by the method of Parnas and Wagner). Z. analyt. Chem. 1938, 114, 261-275.

- Rickli, E. E., S. A. S. Ghazanfar, B. H. Gibbons & J. T. Edsall: Carbonic anhydrase from human erythrocytes. J. biol. Chem. 1964, 239, 1065-1078.
- Rosenberg, A.: The optical rotatory dispersion of bovine and human carbonic anhydrase in the ultraviolet region. J. biol. Chem. 1966, 241, 5126-5136.
- Sandberg, K.: Blood group factors and erythrocytic protein polymorphism in Swedish horses. Proc. 11th Europ. Animal Blood Group Conf., Warsaw 1968, 447-452.
- Sartore, G. & D. Bernoco: Research on biochemical polymorphisms in the indigenous cattle of Piedmont. Proc. 10th Europ. Animal Blood Group Conf., Paris 1966, 283-287.
- Smithies, O.: Zone electrophoresis in starch gels: Group variations in the serum proteins of normal human adults. Biochem. J. 1955, 61, 629-641.
- Spackman, D. H., W. H. Stein & S. Moore: Automatic recording apparatus for use in the chromatography of amino acids. Analyt. Chem. 1958, 30, 1190-1206.
- Spencer, R. L. & F. Wold: A new convenient method for estimation of total cystine-cysteine in proteins. Analyt. Biochem. 1969, 32, 185—190.
- Tanis, R. J., R. E. Tashian & Y.-S. L. Yu: Properties of carbonic anhydrase isozymes isolated from porcine erythrocytes. J. biol. Chem. 1970, 245, 6003-6009.
- Tashian, R. E.: The esterases and carbonic anhydrases of human erythrocytes. In Biochemical Methods in Red Cell Genetics. ed. Yunis, J. J., Acad. Press, New York 1969, p. 307-336.
- Tucker, E. M., Y. Suzuki & C. Stormont: Three new phenotypic systems in the blood of sheep. Vox Sang. (Basel) 1967, 13, 246-262.
- Vesterberg, O. & H. Svensson: Isoelectric fractionation, analysis, and characterization of ampholytes in natural pH gradients. Acta chem. scand. 1966, 20, 820-834.
- Wetlaufer, D. B.: Ultraviolet spectra of proteins and amino acids. Advanc. Protein Chem. 1962, 17, 303-390.
- Wilbur, K. M. & N. G. Anderson: Electrometric and colorimetric determination of carbonic anhydrase. J. biol. Chem. 1948, 176, 147-154.
- Wrigley, C.: Gel electrofocusing a technique for analyzing multiple protein samples by iscelectric focusing. Sci. Tools 1968, 15, 17-23.

SAMMENDRAG

Isolering og karakterisering af tre carboanhydraser fra svineerythrocyter.

Tre distinkte former af det zinkholdige enzym carboanhydrase blev isoleret fra svineerytrocyter. Som fundet hos andre pattedyr indeholder svinets erytrocyter to typer carboanhydrase isoenzymer med forskellige kemiske, fysiske og katalytiske egenskaber. Et lavaktivt enzym med et isoelektrisk pH på 7,15 og et højaktivt enzym med et isoelektrisk pH på 5,97 og en 8 gange højere CO_2 -hydrataseaktivitet isoleredes fra svin af Dansk Landrace. Fra danske sortbrogede svin kunne yderligere isoleres en genetisk variant af det højaktive enzym.

Isoleringsproceduren omfattede en fældning af hæmoglobinet med chloroform-ethanol og separation af isoenzymerne ved DEAE-Sephadex chromatografi. De tre enzymer opførte sig som homogene proteiner ved isoelektrisk focusering og elektroforese ved forskellige pH-værdier. Det lavaktive enzym udgjorde en fjerdedel af det aktive protein i alle chloroform-ethanol ekstrakter. Under fraktioneringen observeredes et stort antal enzymatisk aktive "minor components", som tilsyneladende alle var genetisk koblet med en af de tre hovedkomponenter.

De tre oprensede isoenzymer blev karakteriseret med hensyn til molekylvægt, zinkindhold, aminosyresammensætning, ekstinktionskoefficient og enzymatisk aktivitet overfor CO_2 , p-nitrophenylacetat og β -naphthylacetat. Enzymernes cirkular dichroisme blev også studeret. For de fleste af de studerede egenskabers vedkommende konstateredes tydelig lighed med isoenzymer af tilsvarende type isoleret fra andre pattedyr. Det lavaktive enzym indeholdt ingen svovlholdige aminosyrer. De to højaktive mutanter udviste næsten identisk aminosyresammensætning. Den eneste forskel, som blev observeret, var, at en argininrest i den mest udbredte variant var udskiftet med en histidinrest i den sjældnere variant. Bortset herfra konstateredes ingen signifikante forskelle i de to mutanters egenskaber.

(Received July 6, 1971).

Reprints may be requested from: Jørn Hejgaard, Department of Biochemistry and Nutrition, Technical University of Denmark, 2800 Lyngby, Denmark.